

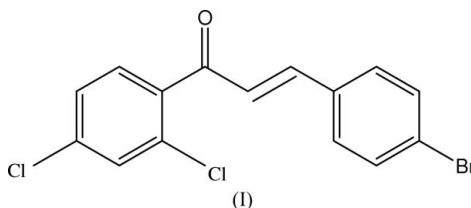
Jeannie Bee-Jan Teh,^a P. S. Patil,^b
Hoong-Kun Fun,^{a*}
Ibrahim Abdul Razak^a and
S. M. Dharmaprasanth^b^aX-ray Crystallography Unit, School of Physics,
Universiti Sains Malaysia, 11800 USM, Penang,
Malaysia, and ^bDepartment of Studies in
Physics, Mangalore University,
Mangalagangothri, Mangalore 574 199, India

Correspondence e-mail: hkfun@usm.my

Key indicators

Single-crystal X-ray study
T = 100 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
R factor = 0.030
wR factor = 0.097
Data-to-parameter ratio = 44.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.3-(4-Bromophenyl)-1-(2,4-dichlorophenyl)-
prop-2-en-1-oneIn the title compound, $\text{C}_{15}\text{H}_9\text{BrCl}_2\text{O}$, the dihedral angle
between the two benzene rings is $42.39(4)^\circ$. The crystal
structure is stabilized by short intermolecular $\text{Br}\cdots\text{Br}$
contacts.Received 4 September 2006
Accepted 5 September 2006

Comment

Our continuing research on NLO materials (Patil *et al.*,
2006*a,b,c*; Chantrapromma *et al.*, 2006) has led us to synthesize
the title compound, (I). We report here the crystal structure of
(I) (Fig. 1). Crystals of (I) do not exhibit second-order non-
linear optical properties as this compound crystallizes in a
centrosymmetric space group.Bond lengths and angles in (I) show normal values (Allen *et al.*, 1987) and are comparable to those in related structures (Teh *et al.*, 2006*a,b*; Patil *et al.*, 2006*a,b,c*). The least-squares plane through the enone unit (C7–C9/O1) makes dihedral angles of $45.42(6)$ and $5.98(11)^\circ$ with the C1–C6 and C10–C15 benzene rings, respectively. The dihedral angle between the two benzene rings is $42.39(4)^\circ$. The relatively short distance [$3.5923(2) \text{ \AA}$] between the Br1 and Br1ⁱ [symmetry code: (i) $1 - x, -y, 1 - z$] atoms indicates the presence of intermolecular $\text{Br}\cdots\text{Br}$ interactions, which contribute to the stabilization of the molecular packing (Fig. 2).

Experimental

2,4-Dichloroacetophenone (0.01 mol) in ethanol (30 ml) was mixed
with 4-bromobenzaldehyde (0.01 mol) in ethanol (30 ml) and the
mixture was treated with an aqueous solution of sodium hydroxide
(5 ml, 20%). This mixture was stirred well and left for 24 h. The
resulting crude solid mass was collected by filtration and recrystallized
from acetone.

Crystal data

 $\text{C}_{15}\text{H}_9\text{BrCl}_2\text{O}$
 $M_r = 356.03$
Monoclinic, $P2_1/c$
 $a = 24.9039(3) \text{ \AA}$
 $b = 3.8474(1) \text{ \AA}$
 $c = 13.9069(2) \text{ \AA}$
 $\beta = 94.923(1)^\circ$
 $V = 1327.58(4) \text{ \AA}^3$ $Z = 4$
 $D_x = 1.781 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
 $\mu = 3.49 \text{ mm}^{-1}$
 $T = 100.0(1) \text{ K}$
Plate, yellow
 $0.21 \times 0.21 \times 0.09 \text{ mm}$

Data collection

Brucker SMART APEX2 CCD
area-detector diffractometer
 ω scans
Absorption correction: multi-scan
(SADABS; Bruker, 2005)
 $T_{\min} = 0.552$, $T_{\max} = 0.738$

42784 measured reflections
7604 independent reflections
5633 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 38.8^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.030$
 $wR(F^2) = 0.098$
 $S = 1.05$
7604 reflections
172 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0553P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.68 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.57 \text{ e } \text{\AA}^{-3}$

H atoms were placed in calculated positions and refined as riding, with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

The authors thank the Malaysian Government and Universiti Sains Malaysia for the Scientific Advancement Grant Allocation (SAGA) grant No.304/PFIZIK/653003/A118. PSP and SMD are grateful to DRDO, Government of India, for financial assistance.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Bruker (2005). APEX2 (Version 1.27), SAINT (Version 7.12A) and SADABS (Version 2004/1). Bruker AXS Inc., Madison, Wisconsin, USA.
Chantrapromma, S., Jindawong, B., Fun, H.-K., Patil, P. S. & Karalai, C. (2006). *Acta Cryst. E62*, o1802–o1804.
Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
Patil, P. S., Teh, J. B.-J., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006a). *Acta Cryst. E62*, o896–o898.
Patil, P. S., Teh, J. B.-J., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006b). *Acta Cryst. E62*, o1710–o1712.
Patil, P. S., Teh, J. B.-J., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006c). *Acta Cryst. E62*, o3096–o3098.

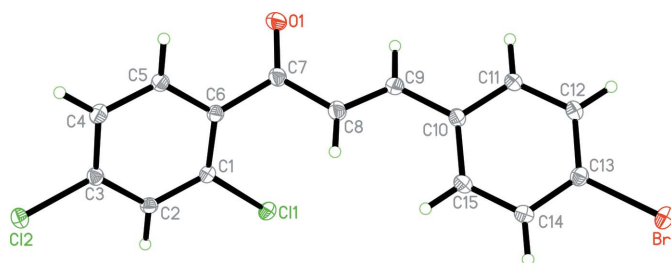


Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atomic numbering.

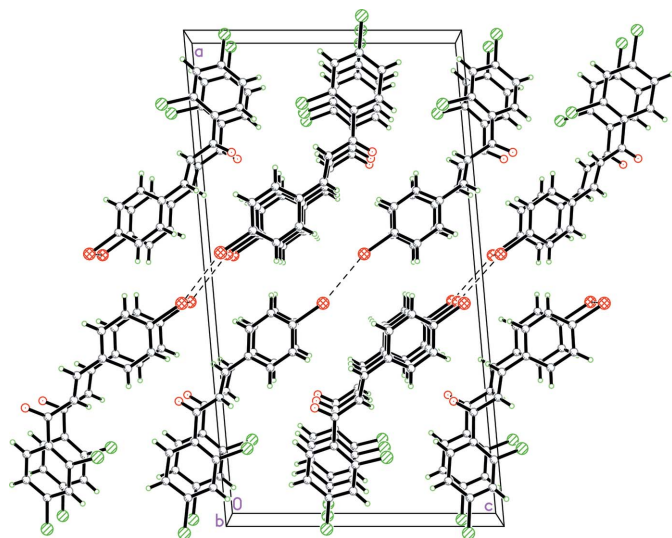


Figure 2

The crystal packing of (I), viewed down the b axis. Dashed lines indicate Br...Br interactions.

- Sheldrick, G. M. (1998). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
Teh, J. B.-J., Patil, P. S., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006a). *Acta Cryst. E62*, o2991–o2992.
Teh, J. B.-J., Patil, P. S., Fun, H.-K., Razak, I. A. & Dharmaparakash, S. M. (2006b). *Acta Cryst. E62*, o2399–o2400.